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Chapter 16 Thermal Energy and Heat Summary 16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects.

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10/28/2014 1 Chapter 16 "Thermal Energy and Heat Section 16.1 "Thermal Energy and Matter " In the 1700s most scientists thought that heat was a fluid called

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Work and Heat Temperature is the measure of how hot or cold something is compared to a reference point. The Celsius scale has reference points of freezing and boiling points of water On the Kelvin scale the reference point is absolute zero Absolute Zero is the temperature at which molecules essentially stop (no kinetic energy)

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heat energy applied to the solid shows up as vibrational energy of the molecules of the solid. The higher the temperature of the solid, the larger is the vibrational motion of its molecules. The lower the temperature, the smaller is the vibrational ... Chapter 16 Temperature and Heat = =

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Chapter 16 Temperature and Heat ... Summary of Chapter 16 " Heat is a form of energy: ... " Energy is conserved in heat flow. Summary of Chapter 16 " Conduction: heat exchange from one part of a material to a cooler part, with no bulk motion of the material.

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concluded that heat is not a form of matter. 474 Chapter 16 FOCUS Objectives 16.1.1 Explain how heat and work transfer energy. 16.1.2 Relate thermal energy to the motion of particles that make ... (Thermal Energy and Matter) at the top. As they move away from the main concept, the content should become more specific.

## **Section 16.1 16.1 Thermal Energy and Matter**

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