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11.2 Types of Chemical Reactions The five general types of reactions are combination, decomposition, single-replacement, double-replacement, and combustion. The number of elements and/or compounds reacting is a good indicator of possible reaction type and thus possible products. In a combination reaction, there is always a single product.

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a) Solve the following problems: Assessment Problems 11.1 – 11.3; Chapter Problems 11.8 – 11.10. 4. Read Section 11.4. a) Show that Eq. (11.21 follows from Eqs. (9.51) – (9.53) on p. 434. b) Derive the line current in terms of the phase currents for a negative phase sequence for a Y- $\hat{I}$  circuit. Follow the example for the positive phase sequence from Eqs. (11.22) – (11.27).

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$v=f\lambda$  The velocity of any wave anywhere is the product of its frequency times its wavelength.  $v$  is velocity, measured in meters per second.  $f$  is frequency, measured in Hertz, which are abbreviated Hz. But 1 Hz is equal to one divided by one second.

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